TUBERCULOSIS CURE XDR=TB

The thermodynamics laws alone determines why only 8 to 10 % of the tuberculosis latent patient gets full TB on set. Weather antibiotic resistant or not.

THE EQUATION IS AS FOLLOWS.

S= E * e raised to tp * log V/N2

S is spreading coefficient, E is entropy, e tp is exponential raised to time and total number of pathogen numbers. Times log V, V is volume of patient. N is the total number of immune cells. This almost looks counter intuitive since it says that as the volume of the carrier latent TB goes down the disease onset reduces ,thus a child may have a lower chance of getting TB and the high mycobacterium tuberculosis still may need correspondingly higher total virus+ bacteria + fungus count. This means a person must have relatively high microorganism count and lower volume and lower immunity cells to get the TB. And thus only low entropy copper has the energy needed to release the dump the energy required to lower the virus+ bacteria and fungus to such a level that the pathogen count starts reducing to barest level. Eventually the body surface like skin surface and buckle cavity and ear glands and eye glands and salivary glands gets lower and lower microorganism count and the entropy starts reducing on the surface to such an extent that the entropy on the surface starts reducing and the microorganism inside the body, where the entropy is still high starts the transportation of microorganism from the blood and the tissue to the surface .slowly the entropy of the body too starts reducing along with all the microorganisms and the transport stops only when the difference between the entropy inside the body and on the outer surface is nearly same. Still the outer surface entropy will be slightly lower than the inside body. The details can be found in the instagram video on from my account 18-8-2026